

Column1	Column2	Column3	Column4	Column42	Column5	Column6
SUBJECT		CODE	DEPARTMENT OF P U EDUCATION			ACADEMIC PROGRAM FOR THE YEAR 2018-19
CHEMISTRY	CLASS	34	<i>PUC (4 THEORY+2 PRACTICE HOURS A WEEK)</i>	PRACTICE SESSIONS		<i>PRACTICALS (1 CLASS OF 2 HOURS DURATION PER WEEK PER BATCH)</i>
DAY	DATE	DAY				
DAY 1	02-May-18	WEDNESDAY	Unit - 1: The Solid State • to describe general characteristics of solid state •to distinguish between amorphous and crystalline solids.			
DAY 2	03-May-18	THURSDAY	•classify crystalline solids on the basis of the nature of binding forces •define crystal lattice and unit cell			
DAY 3	04-May-18	FRIDAY	•to explain close packing of particles •to describe different types of voids and closed packed structures			
DAY 4	05-May-18	SATURDAY	•to calculate the packing efficiency of different types cubic unit cells •to corelate the density of a substance with it's unit cell properties			
DAY 5	06-May-18	SUNDAY				
DAY 6	07-May-18	MONDAY		PRACTICE SESSIONS		
DAY 7	08-May-18	TUESDAY		PRACTICE SESSIONS		
DAY 8	09-May-18	WEDNESDAY	•to describe imperfections in solids and their effect on properties •to corelate the electrical and magnetic properties of solids			
DAY 9	10-May-18	THURSDAY				Surface Chemistry : Preparation of one Lyophillic(Starch sol) and Lyophobic sol(Ferric hydroxide sol) , To purify prepared sol by dialysis
DAY 10	11-May-18	FRIDAY	Election Duty			
DAY 11	12-May-18	SATURDAY	Election Duty			
DAY 12	13-May-18	SUNDAY				
DAY 13	14-May-18	MONDAY	Numericals on formula of a compound and number of voids filled			Chemical Kinetics : Effect of concentration on rate of reaction between Sodium thiosulphate and hydrochloric acid.
DAY 14	15-May-18	TUESDAY	Numericals on density			
DAY 15	16-May-18	WEDNESDAY		PRACTICE SESSIONS		
DAY 16	17-May-18	THURSDAY		PRACTICE SESSIONS		
DAY 17	18-May-18	FRIDAY	Unit - 6: General Principles and processes of Isolation of Elements •to explain the terms minerals, ores, concentration, calcination, roasting, refining, etc. •to understand the principles of oxidation and reduction as applied to the extraction procedures			
DAY 18	19-May-18	SATURDAY	•to apply the thermodynamic concepts like that of Gibbs energy and entropy to the principles of extraction of 1. aluminium 2.copper, 3.zinc.			
DAY 19	20-May-18	SUNDAY				
DAY 20	21-May-18	MONDAY	4.Iron extraction of copper from low grade ores and scrapes •extraction of chlorine from Brime solution(oxidation) •extraction of gold and silver involving leaching with Cyanide ion(CN ⁻)			Effect of temperature on rate of reaction between Sodium thiosulphate and hydrochloric acid.
DAY 21	22-May-18	TUESDAY	•refining techniques: Distillation, Liqueation, Electrolysis, Zone Refining			
DAY 22	23-May-18	WEDNESDAY		PRACTICE SESSIONS		

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CHEMISTRY	CLASS	34			
DAY	DATE	DAY			
DAY 23	24-May-18	THURSDAY	PUC (4 THEORY-2 PRACTICE HOURS A WEEK)	PRACTICE SESSIONS	
DAY 24	25-May-18	FRIDAY	<ul style="list-style-type: none"> Vapour Phase Refining and Chromatographic methods Uses of aluminium, copper, Zinc and Iron 		
DAY 25	26-May-18	SATURDAY	Unit 10: Haloalkanes and Haloarenes <ul style="list-style-type: none"> classification of Haloalkanes and Haloarenes on the basis of <ol style="list-style-type: none"> number of Halogen atoms compounds containing sp^3 C-X bond compounds containing sp^2 C-X bond IUPAC nomenclature of Haloalkanes and Haloarenes 		
DAY 26	27-May-18	SUNDAY			
DAY 27	28-May-18	MONDAY	<ul style="list-style-type: none"> nature of C-X bond methods of preparation <ol style="list-style-type: none"> from alcohols from hydrocarbons by electrophilic substitution sandmayer's reaction from alkenes halogen exchange method 		To determine the heat of solution of potassium nitrate crystals or copper sulphate. To determine the heat of neutralization of strong acid (HCl) with a strong base (NaOH).
DAY 28	29-May-18	TUESDAY	<ul style="list-style-type: none"> physical properties chemical reactions <ol style="list-style-type: none"> Reactions of haloalkanes <ol style="list-style-type: none"> Nucleophilic substitution reaction (with examples) Mechanism of substitution nucleophilic bimolecular (S_N2) 		
DAY 29	30-May-18	WEDNESDAY		PRACTICE SESSIONS	
DAY 30	31-May-18	THURSDAY		PRACTICE SESSIONS	
DAY 31	01-Jun-18	FRIDAY	<ul style="list-style-type: none"> Mechanism of substitution nucleophilic unimolecular (S_N1) Stereochemical aspects of nucleophilic substitution reactions 		
DAY 32	02-Jun-18	SATURDAY	<ul style="list-style-type: none"> Stereochemical aspects of nucleophilic substitution reactions (Contd.) Elimination reactions Reactions with metals <ul style="list-style-type: none"> Preparation of Grignard reagents and its reactivity Wurtz reaction 		
DAY 33	03-Jun-18	SUNDAY			
DAY 34	04-Jun-18	MONDAY	Reactions of Haloarenes <ol style="list-style-type: none"> Nucleophilic substitution reaction to explain why aryl halides are less reactive towards nucleophilic substitution reactions using the following reasons <ol style="list-style-type: none"> Resonance effect Difference in hybridisation of carbon atom C-X bond Instability of phenyl cation Possible repulsion between nucleophile and electron rich arenes 		
DAY 35	05-Jun-18	TUESDAY	<ul style="list-style-type: none"> Electrophilic substitution reactions Polyhalogen compounds 		
DAY 36	06-Jun-18	WEDNESDAY		PRACTICE SESSIONS	
DAY 37	07-Jun-18	THURSDAY		PRACTICE SESSIONS	
DAY 38	08-Jun-18	FRIDAY	Unit 2: Solutions <ul style="list-style-type: none"> Types of solutions expressing concentration of solutions 		

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DAY	DATE	DAY			PRACTICALS (1 CLASS OF 2 HOURS DURATION PER WEEK PER BATCH)
DAY 39	09-Jun-18	SATURDAY	<ul style="list-style-type: none"> •Solubility 1.Solubility of a solid in a liquid, effect of temperature and effect of pressure. 2.Solubility of a gas in a liquid •Henry's Law •Mathematical expression of Henry's law 		
DAY 40	10-Jun-18	SUNDAY			
DAY 41	11-Jun-18	MONDAY	<ul style="list-style-type: none"> •Henry's law constant K_H •Explanation of solubility of the gas in the liquid using K_H •Application of Henry's law in industry •Effect of temperature on the solubility of gases in liquids 		THE DANIELL'S CELL :To set up a Daniell cell and To study the variation of cell potential in $Zn Zn^{2+} Cu^{2+} Cu$ with change in concentration of electrolytes ($CuSO_4$ or $ZnSO_4$) at room temperature.
DAY 42	12-Jun-18	TUESDAY	<ul style="list-style-type: none"> •Vapour pressure of liquid solutions 1. vapour pressure of liquid liquid solutions Raoult's law and mathematical expression •Raoult's law as a special case of Henry's law 2. Vapour pressure of solids in liquids Ideal and non-ideal solutions differences between ideal and non-ideal solutions 		
DAY 43	13-Jun-18	WEDNESDAY		PRACTICE SESSIONS	
DAY 44	14-Jun-18	THURSDAY		PRACTICE SESSIONS	
DAY 45	15-Jun-18	FRIDAY	<ul style="list-style-type: none"> •Azeotropes •Minimum boiling Azeotropes and Maximum boiling Azeotropes •Colligative properties and determination of molar mass 1.Relative lowering of vapour pressure 2.Elevation of boiling point 		
DAY 46	16-Jun-18	SATURDAY	RAMZAN		
DAY 47	17-Jun-18	SUNDAY			
DAY 48	18-Jun-18	MONDAY	<ul style="list-style-type: none"> 3.Depression of freezing point. 4.Osmosis and Osmotic pressure •Isotonic Solutions, hypertonic and hypotonic solutions •Reverse Osmosis and water purification •abnormal molar mass 		
DAY 49	19-Jun-18	TUESDAY	<ul style="list-style-type: none"> •Numericals on concentration of solutions •Numericals on Henry's law 		
DAY 50	20-Jun-18	WEDNESDAY	<ul style="list-style-type: none"> •Numericals on relative lowering of vapour pressure •Numericals on elevation of boiling point 		
DAY 51	21-Jun-18	THURSDAY		PRACTICE SESSIONS	
DAY 52	22-Jun-18	FRIDAY		PRACTICE SESSIONS	
DAY 53	23-Jun-18	SATURDAY	<ul style="list-style-type: none"> •Numericals on depression of freezing point •Numericals on osmotic pressure 		
DAY 54	24-Jun-18	SUNDAY			
DAY 55	25-Jun-18	MONDAY	<ul style="list-style-type: none"> Unit 4 : Chemical Kinetics Introduction, Rate of a chemical reaction (Average and instantaneous) Units of the rate. 		Paper Chromatography :To separate the coloured components present in a mixture of red and blue ink by ascending paper chromatography and find their R_f values.

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DAY	DATE	DAY			
DAY 56	26-Jun-18	TUESDAY	Problems on calculation of average rate. Factors influencing the rate of a reaction		
DAY 57	27-Jun-18	WEDNESDAY	Order of a reaction units for rate constants of zero, first and second order reactions		
DAY 58	28-Jun-18	THURSDAY		PRACTICE SESSIONS	
DAY 59	29-Jun-18	FRIDAY		PRACTICE SESSIONS	
DAY 60	30-Jun-18	SATURDAY	Molecularity of a reaction. Derivation of integrated rate equations for zero and first order reactions.		
DAY 61	01-Jul-18	SUNDAY			
DAY 62	02-Jul-18	MONDAY	Expression for K for first order gas phase reaction. Problems on the above expressions.		Preparation of inorganic compounds : To prepare double salt of ferrous ammonium sulphate or Mohr's salt, Potash alum.
DAY 63	03-Jul-18	TUESDAY	Half life period, derivation of expressions for T _{1/2} of zero and first order reactions. Problems on half life period.		
DAY 64	04-Jul-18	WEDNESDAY	Pseudo first order reaction examples. Temperature dependence of the rate of the reaction.		
DAY 65	05-Jul-18	THURSDAY		PRACTICE SESSIONS	
DAY 66	06-Jul-18	FRIDAY		PRACTICE SESSIONS	
DAY 67	07-Jul-18	SATURDAY	Arrhenius equation , Numericals		
DAY 68	08-Jul-18	SUNDAY			
DAY 69	09-Jul-18	MONDAY	Effect of the temperature on the rate of the reaction. Collision theory		. Preparation of Organic compounds :To prepare a pure sample of dibenzal acetone. To prepare a pure sample of p- nitroacetanilide To prepare a sample of β- naphthol aniline dye (phenyl-azo-β-naphthol)
DAY 70	10-Jul-18	TUESDAY	Unit 11- alcohols, phenols and ethers Alcohols :Classification, nomenclature, structure of functional group		
DAY 71	11-Jul-18	WEDNESDAY	Methods of preparation from alkenes, aldehydes and ketones ones, carboxylic acid, grignard reagent Physical Properties - Boiling point and solubility		
DAY 72	12-Jul-18	THURSDAY		PRACTICE SESSIONS	
DAY 73	13-Jul-18	FRIDAY		PRACTICE SESSIONS	
DAY 74	14-Jul-18	SATURDAY	Chemical properties - Acidity, Esterification, acylation, oxidation		
DAY 75	15-Jul-18	SUNDAY			
DAY 76	16-Jul-18	MONDAY	Dehydration of alcohols, mechanism of dehydration, De-hydrogenation, Lucas reagent test, Manufacture of methanol and ethanol, Uses of methanol and ethanol		Test for the Functional Groups Present in Organic Compounds
DAY 77	17-Jul-18	TUESDAY	Phenols - Classification, nomenclature, preparation		
DAY 78	18-Jul-18	WEDNESDAY	Physical properties, Chemical properties - Acidity, Esterification.		
DAY 79	19-Jul-18	THURSDAY			
DAY 80	20-Jul-18	FRIDAY			
DAY 81	21-Jul-18	SATURDAY			
DAY 82	22-Jul-18	SUNDAY			
DAY 83	23-Jul-18	MONDAY		PRACTICE SESSIONS	Study of Carbohydrates , Fats and Protiens in Pure Form and Detection of their Presence in Given Food Stuffs.
DAY 84	24-Jul-18	TUESDAY		PRACTICE SESSIONS	

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DAY 85	25-Jul-18	WEDNESDAY	Nitration, Halogenation of Phenol, Kolbe's reaction, Reimer Tiemann reaction, Phenol with Zinc dust and oxidation of Phenol	
DAY 86	26-Jul-18	THURSDAY	Ethers	
DAY 87	27-Jul-18	FRIDAY	Unit 12: Aldehydes, Ketones and Carboxylic acids Introduction, Nomenclature of aldehydes and ketones Nature of carbonyl group	
DAY 88	28-Jul-18	SATURDAY	Preparation of aldehydes and ketones	
DAY 89	29-Jul-18	SUNDAY		
DAY 90	30-Jul-18	MONDAY		PRACTICE SESSIONS
DAY 91	31-Jul-18	TUESDAY		PRACTICE SESSIONS
DAY 92	01-Aug-18	WEDNESDAY	Properties - Physical Properties, Chemical Properties, addition reactions Mechanism of addition of HCN	
DAY 93	02-Aug-18	THURSDAY	Condensation reactions, Clemmensen reduction, Wolff-Kishner reduction	
DAY 94	03-Aug-18	FRIDAY	Tests to distinguish aldehydes and ketones - Haloform reaction, Aldol condensation	
DAY 95	04-Aug-18	SATURDAY	Cannizzaro's reaction, Uses of aldehydes and ketones, Carboxylic acids - Nomenclature	
DAY 96	05-Aug-18	SUNDAY		
DAY 97	06-Aug-18	MONDAY		PRACTICE SESSIONS
DAY 98	07-Aug-18	TUESDAY		PRACTICE SESSIONS
DAY 99	08-Aug-18	WEDNESDAY	Structure of carboxylic group, methods of preparation of carboxylic acids	
DAY 100	09-Aug-18	THURSDAY	Physical properties and reactions of carboxylic acids	
DAY 101	10-Aug-18	FRIDAY	HVZ reaction and electrophilic reactions and uses of carboxylic acids	
DAY 102	11-Aug-18	SATURDAY	Unit 7: P-Block elements Nitrogen family - occurrence, electronic configuration, oxidation state, atomic and ionic radii, ionisation energy, electro negativity	
DAY 103	12-Aug-18	SUNDAY		
DAY 104	13-Aug-18	MONDAY		PRACTICE SESSIONS
2E	14-Aug-18	TUESDAY		PRACTICE SESSIONS
DAY 106	15-Aug-18	WEDNESDAY	INDEPENDENCE DAY	
DAY 107	16-Aug-18	THURSDAY	Chemical properties of P block elements, preparation of di-nitrogen, properties and uses of di-nitrogen	
DAY 108	17-Aug-18	FRIDAY	Ammonia - Preparation, manufacture by Haber's process, properties and uses	
DAY 109	18-Aug-18	SATURDAY	Oxides of nitrogen - Methods of preparation, structure, appearance and chemical nature Nitric acid - Manufacture by Ostwald's process	
DAY 110	19-Aug-18	SUNDAY		
DAY 111	20-Aug-18	MONDAY	Properties and uses of Nitric acid Phosphorus - Allotropic forms, Preparation, properties and uses of Phosphine	
DAY 112	21-Aug-18	TUESDAY		PRACTICE SESSIONS
DAY 113	22-Aug-18	WEDNESDAY	BAKRID	

. Determination of Concentration / Molarity of $KMnO_4$ solution by Titrating it against a standard solution of Oxalic Acid

. Determination of Concentration / Molarity of $KMnO_4$ solution by Titrating it against a standard solution of Ferruous Ammonium Sulphate.

Qualitative Analysis : Determination of one cation and one anion in a given salt.

Qualitative Analysis : Determination of one cation and one anion in a given salt.

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DAY 114	23-Aug-18	THURSDAY			
DAY 115	24-Aug-18	FRIDAY	Phosphorus Halides - Preparation, properties and structure Oxoacids of phosphorus Group 16 elements - Occurrence, electronic configuration, electron gain enthalpy and other physical properties		
DAY 116	25-Aug-18	SATURDAY	Chemical properties of Group 16 elements Di-oxygen - preparation, properties and uses Ozone - preparation, properties and uses		
DAY 117	26-Aug-18	SUNDAY			
DAY 118	27-Aug-18	MONDAY	Sulphur - Allotropic forms Sulphur-di-oxide - preparation, properties and uses Oxoacids of sulphur		
DAY 119	28-Aug-18	TUESDAY	Sulphuric acid - manufacture by contact process, properties, uses. Group 17 elements physical and chemical properties		
DAY 120	29-Aug-18	WEDNESDAY			PRACTICE SESSIONS
DAY 121	30-Aug-18	THURSDAY			PRACTICE SESSIONS
DAY 122	31-Aug-18	FRIDAY	Chlorine preparation, properties and uses. Hydrogen chloride preparation, properties and uses		
DAY 123	01-Sep-18	SATURDAY	Oxo Acids of Halogens, inter halogen compounds. Group 18 elements		
DAY 124	02-Sep-18	SUNDAY			
DAY 125	03-Sep-18	MONDAY	UNIT 5 : SURFACE CHEMISTRY : Adsorption: adsorbate, adsorbent, examples, distinction between adsorption and absorption. H, S and G for adsorption of gas on a solid. Physisorption and chemisorption-characteristics and differences. Factors affecting adsorption of a gas on a solid		Qualitative Analysis : Determination of one cation and one anion in a given salt.
DAY 126	04-Sep-18	TUESDAY	. Applications of adsorption . Catalysis: homogeneous and heterogeneous catalysis, examples, activity and selectivity of a catalyst, examples, shape selective catalysis, examples. Enzyme catalysis: examples, characteristics (to be mentioned), mechanism.		
DAY 127	05-Sep-18	WEDNESDAY			PRACTICE SESSIONS
DAY 128	06-Sep-18	THURSDAY			PRACTICE SESSIONS
DAY 129	07-Sep-18	FRIDAY	Colloids: colloidal state-distinction of true solution, colloids, and suspension based on particle size. Classification of colloids-types of colloidal systems-examples, lyophilic and lyophobic— differences and examples, macromolecular, multimolecular and associated colloids, examples		
DAY 130	08-Sep-18	SATURDAY	formation of micelle, cleansing action of soaps. Preparation of colloids-chemical methodssulphur and ferric hydroxide sols, Bredig's arc method for metal sols, peptisation		
DAY 131	09-Sep-18	SUNDAY			
DAY 132	10-Sep-18	MONDAY			
DAY 133	11-Sep-18	TUESDAY			
DAY 134	12-Sep-18	WEDNESDAY			
DAY 135	13-Sep-18	THURSDAY	GANESH CHATURTHI		
DAY 136	14-Sep-18	FRIDAY			
DAY 137	15-Sep-18	SATURDAY			
DAY 138	16-Sep-18	SUNDAY			
DAY 139	17-Sep-18	MONDAY			
DAY 140	18-Sep-18	TUESDAY			
DAY 141	19-Sep-18	WEDNESDAY			

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DAY	DATE	DAY		PRACTICALS (1 CLASS OF 2 HOURS DURATION PER WEEK PER BATCH)	
DAY 142	20-Sep-18	THURSDAY			
DAY 143	21-Sep-18	FRIDAY	LAST DAY OF MOHARRUM		
DAY 144	22-Sep-18	SATURDAY	Purification– dialysis, electro-dialysis, ultrafiltration (in brief). Properties of colloids: Tyndall effect, Brownian movement, charge on colloidal particles, examples, electrophoresis,		
DAY 145	23-Sep-18	SUNDAY			
DAY 146	24-Sep-18	MONDAY	Coagulation – methods of coagulation of lyophobic sols, Hardy- Schulze rule-examples, coagulating value. Protective colloid - example. Applications: In industries, medicines, purification of drinking water. Emulsions : types , examples		Qualitative Analysis : Determination of one cation and one anion in a given salt.
DAY 147	25-Sep-18	TUESDAY		PRACTICE SESSIONS	
DAY 148	26-Sep-18	WEDNESDAY		PRACTICE SESSIONS	
DAY 149	27-Sep-18	THURSDAY	UNIT-III Electrochemistry -Redox reaction – As fundamental reaction in electrochemical cells, electronic and electrolytic conductors – differences, strong and weak electrolytes, examples-		
DAY 150	28-Sep-18	FRIDAY	Ionic conductance- factors affecting ionic conductance, conductivity and molar conductivity of electrolytic solutions- definitions, mathematical expressions, relationship between them, SI units, numerical problems.		
DAY 151	29-Sep-18	SATURDAY	Variation of conductivity and molar conductivity with concentration, graph for variation of Λ_m vs $C^{1/2}$ for strong and weak electrolytes using equation $\Lambda_m = \Lambda_m^\infty - A C^{1/2}$ (measurement of conductivity from Wheatstone network not included), limiting molar conductivities, Kohlrausch law and applications,		
DAY 152	30-Sep-18	SUNDAY			
DAY 153	01-Oct-18	MONDAY	Numerical problems on calculation of Λ_m^∞ for weak electrolytes. Electrolysis –Faraday's laws of electrolysis (elementary idea) , concept of nF required to discharge one mole of M^{n+} ions, numerical problems on I law. Galvanic cells : Electrode potential		Qualitative Analysis : Determination of one cation and one anion in a given salt.
DAY 154	02-Oct-18	TUESDAY	MAHATHMA GANDHI JAYANTHI		
DAY 155	03-Oct-18	WEDNESDAY		PRACTICE SESSIONS	
DAY 156	04-Oct-18	THURSDAY		PRACTICE SESSIONS	
DAY 157	05-Oct-18	FRIDAY	Half cell concept, standard electrode potential, galvanic cell, Daniell cell, cell potential, EMF (emf), $E_{cell}^\ominus = E_{right}^\ominus - E_{left}^\ominus$ Measurement of electrode potential – SHE - diagram, half cell representation, half cell reaction, E^\ominus taken as 0.0 V (at all temperatures).		
DAY 158	06-Oct-18	SATURDAY	Measurement of E^\ominus of Zn and Cu using SHE (experimental details not expected) numerical problems on E^\ominus Importance of standard electrode potentials- to decide and compare the strengths of oxidising and reducing agents .Nernst equation (derivation not required) : Nernst equation at 298 K for single electrode potential and cell potential,		
DAY 159	07-Oct-18	SUNDAY			
DAY 160	08-Oct-18	MONDAY	MAHALAYA IMMAYASAYA		
DAY 161	09-Oct-18	TUESDAY	Numerical problems to calculate half cell and cell potentials (only for metal electrodes). Relationship between equilibrium constant.Relationship between equilibrium constant and E_{cell}^\ominus (derivation not required), numerical problems. Relationship between standard Gibbs energy and E_{cell}^\ominus ,numerical problems.		Qualitative Analysis : Determination of one cation and one anion in a given salt.

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DAY 162	10-Oct-18	WEDNESDAY	Factors affecting the products of electrolysis, examples – molten and aqueous solution of NaCl only. Batteries: types-difference, examples, Leclanche cell (dry cell)	
DAY 163	11-Oct-18	THURSDAY		PRACTICE SESSIONS
DAY 164	12-Oct-18	FRIDAY		PRACTICE SESSIONS
DAY 165	13-Oct-18	SATURDAY	Lead acid battery–anode, cathode, electrolyte, reactions at anode and cathode (diagram not required), Fuel cell – definition – examples, H ₂ -O ₂ fuel cell – schematic diagram, anode, cathode, electrolyte, reactions at anode and cathode. Corrosion – rusting of iron- anodic, cathodic reactions, composition of rust, methods of prevention	
DAY 166	14-Oct-18	SUNDAY		
DAY 167	15-Oct-18	MONDAY		
DAY 168	16-Oct-18	TUESDAY		
DAY 169	17-Oct-18	WEDNESDAY		
DAY 170	18-Oct-18	THURSDAY	MAHANAVAMI	
DAY 171	19-Oct-18	FRIDAY	VIJAYADASHMI	
DAY 172	20-Oct-18	SATURDAY		
DAY 173	21-Oct-18	SUNDAY		
DAY 174	22-Oct-18	MONDAY		
DAY 175	23-Oct-18	TUESDAY		
DAY 176	24-Oct-18	WEDNESDAY	VALMIKI JAYANTHI	
DAY 177	25-Oct-18	THURSDAY		
DAY 178	26-Oct-18	FRIDAY		
DAY 179	27-Oct-18	SATURDAY		
DAY 180	28-Oct-18	SUNDAY		
DAY 181	29-Oct-18	MONDAY	UNIT VIII d and f Block Elements General introduction, electronic configuration, characteristics of transition metals (d-block) - variation in atomic and ionic size.	
DAY 182	30-Oct-18	TUESDAY	Electronic configuration of 3d series elements, general trends in properties of the first row transition metals (3d series) – metallic character, ionization enthalpies, oxidation states	Qualitative Analysis : Determination of one cation and one anion in a given salt.
DAY 183	31-Oct-18	WEDNESDAY	Magnetic properties, colour, catalytic properties, formation of interstitial compounds, alloy formation.	
DAY 184	01-Nov-18	THURSDAY	KANNADA RAJYOTSAVA	
DAY 185	02-Nov-18	FRIDAY		PRACTICE SESSIONS
DAY 186	03-Nov-18	SATURDAY		PRACTICE SESSIONS
DAY 187	04-Nov-18	SUNDAY		
DAY 188	05-Nov-18	MONDAY	Potassium dichromate: preparation from chromite ore (FeCr ₂ O ₄). Properties – oxidizing property – with I ⁻ , H ₂ S, Sn ²⁺ , Fe ²⁺ etc.	Qualitative Analysis : Determination of one cation and one anion in a given salt.
DAY 189	06-Nov-18	TUESDAY	NARAKA CHATURDASHI	
DAY 190	07-Nov-18	WEDNESDAY	Interconversion of chromates and dichromates in aqueous solution depending on pH. Potassium permanganate: Preparation from MnO ₂ by reaction with KOH and acidification.	
DAY 191	08-Nov-18	THURSDAY	BALIPADYAMI DEEPAWALI	
DAY 192	09-Nov-18	FRIDAY	Properties of potassium permanganate – Action of heat, oxidising property-oxidation of I ⁻ , Fe ²⁺ , C ₂ O ₄ ²⁻ , H ₂ S etc. In acidic medium, S ₂ O ₃ ²⁻ , I ⁻ etc in neutral / alkaline medium	
DAY 193	10-Nov-18	SATURDAY	f-block elements: Lanthanoids-electronic configuration, atomic size- lanthanoid contraction and its consequences	

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VACATION

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DAY 194	11-Nov-18	SUNDAY		
DAY 195	12-Nov-18	MONDAY		Qualitative Analysis : Determination of one cation and one anion in a given salt.
DAY 196	13-Nov-18	TUESDAY		
DAY 197	14-Nov-18	WEDNESDAY	Oxidation states of f – block elements , chemical reactivity –general characteristics. Actinoids; electronic configuration	
DAY 198	15-Nov-18	THURSDAY	Ionic size – actinoid contraction – compared to lanthanoid contraction, oxidation states– general characteristics compared with lanthanoids.	
DAY 199	16-Nov-18	FRIDAY	Unit 13: Amines. Structure of amines. Classification. Nomenclature of amines.	
DAY 200	17-Nov-18	SATURDAY	Preparation of amines. Physical properties.	
DAY 201	18-Nov-18	SUNDAY		
DAY 202	19-Nov-18	MONDAY		PRACTICE SESSIONS
DAY 203	20-Nov-18	TUESDAY		Qualitative Analysis : Determination of one cation and one anion in a given salt.
DAY 204	21-Nov-18	WEDNESDAY	EID MILAD	
DAY 205	22-Nov-18	THURSDAY	Basic character of amines. Structure-basicity relationship of amines.	
DAY 206	23-Nov-18	FRIDAY	Chemical reactions of amines	
DAY 207	24-Nov-18	SATURDAY	Diazonium salts. Methods of preparation. Physical properties.	
DAY 208	25-Nov-18	SUNDAY		
DAY 209	26-Nov-18	MONDAY	KANAKLDAS JAYANTHI	
DAY 210	27-Nov-18	TUESDAY	Chemical reactions of diazonium salts. Importance of diazonium salts in synthesis of aromatic compounds.	Qualitative Analysis : Determination of one cation and one anion in a given salt.
DAY 211	28-Nov-18	WEDNESDAY		PRACTICE SESSIONS
DAY 212	29-Nov-18	THURSDAY		PRACTICE SESSIONS
DAY 213	30-Nov-18	FRIDAY	Unit 9: Coordination compounds. Werner's theory of coordination compounds. Difference between a double salt and a complex	
DAY 214	01-Dec-18	SATURDAY	Definitions of some important terms pertaining to coordination compounds.	
DAY 215	02-Dec-18	SUNDAY		
DAY 216	03-Dec-18	MONDAY	Nomenclature of Coordination compounds.	
DAY 217	04-Dec-18	TUESDAY	Isomerism in coordination compounds. Stereoisomerism and structural isomerism	
DAY 218	05-Dec-18	WEDNESDAY		PRACTICE SESSIONS
DAY 219	06-Dec-18	THURSDAY		
DAY 220	07-Dec-18	FRIDAY		
DAY 221	08-Dec-18	SATURDAY		
DAY 222	09-Dec-18	SUNDAY		
DAY 223	10-Dec-18	MONDAY		PRACTICE SESSIONS
DAY 224	11-Dec-18	TUESDAY	Bonding in coordination compounds. Valence bond theory and its limitations.	Qualitative Analysis : Determination of one cation and one anion in a given salt.
DAY 225	12-Dec-18	WEDNESDAY	Crystal field theory. Colour in coordination compounds. Limitations of crystal field theory.	
DAY 226	13-Dec-18	THURSDAY	Bonding in metal carbonyls. Stability of coordination compounds. Importance and applications of coordination compounds.	
DAY 227	14-Dec-18	FRIDAY	Unit 14: Biomolecules. Carbohydrates. Classification of carbohydrates. Monosaccharides. Preparation of glucose.	
DAY 228	15-Dec-18	SATURDAY		PRACTICE SESSIONS
DAY 229	16-Dec-18	SUNDAY		

2 TEST

			DEPARTMENT OF P U EDUCATION		ACADEMIC PROGRAM FOR THE YEAR 2018-19		
SUBJECT	CODE						
CHEMISTRY	CLASS	34	<i>PUC (4 THEORY-2 PRACTICE HOURS A WEEK)</i>			PRACTICE SESSIONS	
						<i>PRACTICALS (1 CLASS OF 2 HOURS DURATION PER WEEK PER BATCH)</i>	
DAY	DATE	DAY					
DAY 230	17-Dec-18	MONDAY					Qualitative Analysis : Determination of one cation and one anion in a given salt.
DAY 231	18-Dec-18	TUESDAY	Structure of glucose. Cyclic structure of glucose. Structure of fructose.				
DAY 232	19-Dec-18	WEDNESDAY	Disaccharides, polysaccharides, importance of carbohydrates.				
DAY 233	20-Dec-18	THURSDAY	Proteins. Amino acids. Classification of amino acids.				
DAY 234	21-Dec-18	FRIDAY	Peptides, structure of proteins. Denaturation of proteins.				
DAY 235	22-Dec-18	SATURDAY					PRACTICE SESSIONS
DAY 236	23-Dec-18	SUNDAY					
DAY 237	24-Dec-18	MONDAY					PRACTICE SESSIONS
DAY 238	25-Dec-18	TUESDAY	CHRISTMAS				Qualitative Analysis : Determination of one cation and one anion in a given salt.
DAY 239	26-Dec-18	WEDNESDAY	Enzymes. Vitamins. Classification of vitamins. Nucleic acids.				
DAY 240	27-Dec-18	THURSDAY	Structure of Nucleic acids. Biological functions of Nucleic acids.				
DAY 241	28-Dec-18	FRIDAY	Unit 15: Polymers. Polymer. Polymerisation. Classification of Polymers based on source and structure.				
DAY 242	29-Dec-18	SATURDAY	Classification of polymers based on mode of polymerisation and molecular force				
DAY 243	30-Dec-18	SUNDAY					
DAY 244	31-Dec-18	MONDAY					PRACTICE SESSIONS
DAY 245	01-Jan-19	TUESDAY					Qualitative Analysis : Determination of one cation and one anion in a given salt.
DAY 246	02-Jan-19	WEDNESDAY	Types of polymerisation reactions. Addition polymerisation. Preparation of some important addition polymers.				
DAY 247	03-Jan-19	THURSDAY	Condensation polymerisation and copolymers. (including examples)				
DAY 248	04-Jan-19	FRIDAY	Rubber. Molecular mass of polymers. Biodegradable polymers.				
DAY 249	05-Jan-19	SATURDAY	Unit 16: Chemistry in Everyday life. Drugs and their classification. Drug - target interaction				
DAY 250	06-Jan-19	SUNDAY					
DAY 251	07-Jan-19	MONDAY					
DAY 252	08-Jan-19	TUESDAY					
DAY 253	09-Jan-19	WEDNESDAY					
DAY 254	10-Jan-19	THURSDAY					
DAY 255	11-Jan-19	FRIDAY					
DAY 256	12-Jan-19	SATURDAY					
DAY 257	13-Jan-19	SUNDAY					
DAY 258	14-Jan-19	MONDAY					
DAY 259	15-Jan-19	TUESDAY					
DAY 260	16-Jan-19	WEDNESDAY					
DAY 261	17-Jan-19	THURSDAY					
DAY 262	18-Jan-19	FRIDAY					PRACTICE SESSIONS
DAY 263	19-Jan-19	SATURDAY					PRACTICE SESSIONS
DAY 264	20-Jan-19	SUNDAY					
DAY 265	21-Jan-19	MONDAY	Therapeutic action of different classes of drugs. Antacids. Antihistamines.				
DAY 266	22-Jan-19	TUESDAY	Neurologically active drugs. Antimicrobials. Antifertility drugs.				
DAY 267	23-Jan-19	WEDNESDAY	Chemicals in food. Artificial sweetening agents. Food preservatives.				
DAY 268	24-Jan-19	THURSDAY	Cleansing agents. Soaps. Types of soaps. Synthetic detergents. Types and uses.				

2NDPUC PREPARATORY EXAM

SUBJECT		CODE	DEPARTMENT OF P U EDUCATION	ACADEMIC PROGRAM FOR THE YEAR 2018-19
CHEMISTRY	CLASS		PUC (4 THEORY-2 PRACTICE HOURS A WEEK)	PRACTICE SESSIONS
		34		PRACTICALS (1 CLASS OF 2 HOURS DURATION PER WEEK PER BATCH)
DAY	DATE	DAY		
DAY 269	25-Jan-19	FRIDAY		PRACTICE SESSIONS
DAY 270	26-Jan-19	SATURDAY		PRACTICE SESSIONS
DAY 271	27-Jan-19	SUNDAY		
DAY 272	28-Jan-19	MONDAY		
DAY 273	29-Jan-19	TUESDAY		
DAY 274	30-Jan-19	WEDNESDAY		
DAY 275	31-Jan-19	THURSDAY		
DAY 276	01-Feb-19	FRIDAY		
DAY 277	02-Feb-19	SATURDAY		
DAY 278	03-Feb-19	SUNDAY		
DAY 279	04-Feb-19	MONDAY		
DAY 280	05-Feb-19	TUESDAY		
DAY 281	06-Feb-19	WEDNESDAY		
DAY 282	07-Feb-19	THURSDAY		
DAY 283	08-Feb-19	FRIDAY		
DAY 284	09-Feb-19	SATURDAY		
DAY 285	10-Feb-19	SUNDAY		
DAY 286	11-Feb-19	MONDAY		
DAY 287	12-Feb-19	TUESDAY		
DAY 288	13-Feb-19	WEDNESDAY		
DAY 289	14-Feb-19	THURSDAY		
DAY 290	15-Feb-19	FRIDAY		
DAY 291	16-Feb-19	SATURDAY		
DAY 292	17-Feb-19	SUNDAY		
DAY 293	18-Feb-19	MONDAY		
DAY 294	19-Feb-19	TUESDAY		
DAY 295	20-Feb-19	WEDNESDAY		
DAY 296	21-Feb-19	THURSDAY		
DAY 297	22-Feb-19	FRIDAY		
DAY 298	23-Feb-19	SATURDAY		
DAY 299	24-Feb-19	SUNDAY		
DAY 300	25-Feb-19	MONDAY		
DAY 301	26-Feb-19	TUESDAY		
DAY 302	27-Feb-19	WEDNESDAY		
DAY 303	28-Feb-19	THURSDAY		
DAY 304	01-Mar-19	FRIDAY		
DAY 305	02-Mar-19	SATURDAY		
DAY 306	03-Mar-19	SUNDAY		
DAY 307	04-Mar-19	MONDAY		
DAY 308	05-Mar-19	TUESDAY		
DAY 309	06-Mar-19	WEDNESDAY		
DAY 310	07-Mar-19	THURSDAY		
DAY 311	08-Mar-19	FRIDAY		
DAY 312	09-Mar-19	SATURDAY		
DAY 313	10-Mar-19	SUNDAY		
DAY 314	11-Mar-19	MONDAY		
DAY 315	12-Mar-19	TUESDAY		
DAY 316	13-Mar-19	WEDNESDAY		
DAY 317	14-Mar-19	THURSDAY		
DAY 318	15-Mar-19	FRIDAY		
DAY 319	16-Mar-19	SATURDAY		
DAY 320	17-Mar-19	SUNDAY		

IPU ANNUAL EXAM

SUBJECT		CODE	DEPARTMENT OF P U EDUCATION		ACADEMIC PROGRAM FOR THE YEAR 2018-19
CHEMISTRY	CLASS	34	PUC (4 THEORY-2 PRACTICE HOURS A WEEK)	PRACTICE SESSIONS	PRACTICALS (1 CLASS OF 2 HOURS DURATION PER WEEK PER BATCH)
DAY	DATE	DAY			
DAY 321	18-Mar-19	MONDAY			
DAY 322	19-Mar-19	TUESDAY			
DAY 323	20-Mar-19	WEDNESDAY			